

## Formation of ZnO films on SiC/porous Si/Si substrates

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**Abstract.** Reactive magnetron sputtering was used to obtain ZnO films on SiC/porous Si/Si substrates. The silicon carbide film on the surface of porous Si was obtained using chemical substitution of atoms. It has been shown that ZnO films grown at the partial oxygen pressure 0.6 Pa are characterized by a smoother and more uniform surface than coatings grown at the oxygen pressure 0.1 Pa. Being based on the analysis of Raman and photoluminescence spectra, it has been shown that the increase in partial pressure of oxygen leads to the increase in structural disorder of the ZnO crystal lattice, on the one hand, and at the same time to a decrease in the concentration of intrinsic defects, including ionized oxygen vacancies  $O_i$ , on the other hand.

**Keywords:** reactive magnetron sputtering, porous Si, silicon carbide film, chemical substitution of atoms, Raman and photoluminescence spectra.

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### 1. Introduction

Zinc oxide (ZnO) is one of the most attractive materials for use in optoelectronic devices due to its wide band gap (3.37 eV) and high exciton binding energy (60 meV) [1–3]. In particular, ZnO is considered as a candidate material for use in optoelectronic devices such as ultraviolet (UV) or blue light-emitting diodes (LED) and laser diodes (LD) [4–8]. However, high-quality ZnO single crystal films are required to be used in manufacturing the high-performance LEDs. Since single crystal ZnO wafers are too expensive, most of ZnO films are produced using heteroepitaxy on foreign substrates.

The Si substrate is a very attractive material for ZnO heteroepitaxy due not only to its low cost, but also to additional advantages in the further integration of the resulting structures into photovoltaic devices. However, it is difficult to obtain high-quality monocrystalline ZnO films on Si substrates because of the large lattice mismatch between ZnO, which reaches approximately 20%; moreover, the crystal lattices of these materials are characterized by different symmetry [9]. Silicon is a cubic crystal, while ZnO has a hexagonal lattice.

The above drawbacks can be eliminated using a buffer layer with suitable structural and thermal properties in fabrication of the ZnO/Si structure. Among

these materials, silicon carbide SiC is a suitable candidate. This semiconductor is attractive due to the similarity of crystal structures with ZnO and small lattice and thermal mismatch, as well as high thermal conductivity, excellent mechanical properties and good chemical stability [10]. The good lattice matching of ZnO and SiC heteroepitaxial materials allows formation of the compounds with excellent electrical and optical properties. In this case, a decrease of the compression strain in the ZnO film will also lead to a decrease in the crystallite size [11]. However, the commercial cost of manufacturing of single crystal SiC films is also very high. The solution of this problem was creation of single-crystal SiC films with a low dislocation density on the surface of Si wafers. The Si substrate can be successfully used to further reduce the cost of manufacturing large area SiC films.

It is also an advantage that when using Si as substrate, for example for fabrication of GaN- or ZnO-based LEDs, the films can be compatible and integrated to form all Si-based optoelectronic devices. To obtain SiC films on single-crystal silicon substrates, a nanoporous Si layer can be used, which plays the role of a soft substrate and adjusts to the structure of the SiC film, reducing mechanical stresses caused by the lattice mismatch between SiC and Si [12].

A silicon carbide film can be obtained by chemical substitution of atoms [12, 13], which enables to successfully combine the advantages of both silicon carbide and silicon, and to obtain thin films of monocrystalline SiC on Si substrates in relatively low-temperature processes (1200...1350 °C) that do not require large energy costs. A distinctive feature of this method is that the silicon carbide film is formed directly in the near-surface bulk of the substrate and from its material – silicon – by substituting some of the Si atoms for C [13].

This work is devoted to the study of properties inherent to the zinc oxide films prepared using magnetron sputtering on a SiC/porous Si/Si template.

## 2. Experimental technique

Micro Raman scattering (Raman) and photoluminescence (PL) spectra of ZnO were recorded using a Horiba Jobin Yvon T64000 triple spectrometer with an Olympus BX41 confocal microscope (×50 objective, aperture 0.75). Micro Raman spectra were measured in backscattering geometry. To excite the Raman spectra, radiation from a Spectra Physics Excelsior DPSS laser with the wavelength  $\lambda_{ex} = 532.0$  nm was used; for excitation of the PL spectra, radiation from a He–Cd laser with the wavelength  $\lambda_{ex} = 325.0$  nm was used. SEM images of the ZnO/SiC/porous Si/Si structures were obtained using a Tescan Mira 3 LMU scanning electron microscope.

### Method for obtaining the ZnO/SiC/porous Si/Si structure

Monocrystalline silicon plates were used as a base for the SiC/porous Si/Si multilayer substrate. The plate diameter was 100 mm, thickness –  $460 \pm 25$  μm, electrical resistivity – 1...10 Ohm·cm, operating side of plates was polished, reverse side was grinding-etched, dislocation density  $<100$  cm<sup>-2</sup>.

The process of electrochemical etching of Si(100) plates for obtaining a porous Si layer consisted of the following three stages.

The stage 1 – pre-cleaning. Before the etching procedure, the following steps were performed to prepare monocrystalline Si(100) plates:

- grinding the samples with diamond paste;
- cleaning of plates with toluene, ethanol and isopropanol;
- washing in running deionized water (removal of pretreatment reaction products);
- drying of the plates by using a centrifuge in a jet of purified dry air.

The stage 2 – actual etching process.

The electrolyte HF:H<sub>2</sub>O:C<sub>2</sub>H<sub>5</sub>OH = 2:1:1 (40% hydrofluoric acid, 96% alcohol) was used in the experiment. A platinum plate with the dimensions 5×10×1 mm was used as a cathode, which was located 2...3 cm away from the semiconductor electrode (anode). The current density was 50 mA/cm<sup>2</sup>, and the duration of the etching process – 1200 s. A DC power supply (30 V, 10 A) was used as a power source.

The stage 3 – cleaning of the processed plates from etching products. After electrochemical treatment, the obtained samples were dried in a flow of a mixture of hydrogen, nitrogen, and hot air.

The SiC film was deposited on a porous Si/Si substrate by using the atom substitution method [12, 13]. Directly before annealing, the air from the chamber was evacuated using a 2NVR-5DM rotary pump to a pressure of 10<sup>-2</sup> to 10<sup>-3</sup> Pa. The porous Si/Si sample placed into the setup was heated by a graphite heater to a temperature of 950 to 1350 °C. Then, using the HORIBA gas regulator, CO and SiH<sub>4</sub> gases were pumped through the connecting gas pipes, the total pressure of the CO + SiH<sub>4</sub> gas mixture was 133 Pa, CO and SiH<sub>4</sub> gas flows were 14 and 3.5 cm<sup>3</sup>/min, respectively. The total pressure in the reaction zone was maintained within 20...600 Pa. The duration of the synthesis process was 20...60 min.

Technological conditions for the synthesis of SiC layers on samples of porous-Si/Si (100) are listed in Table 1.

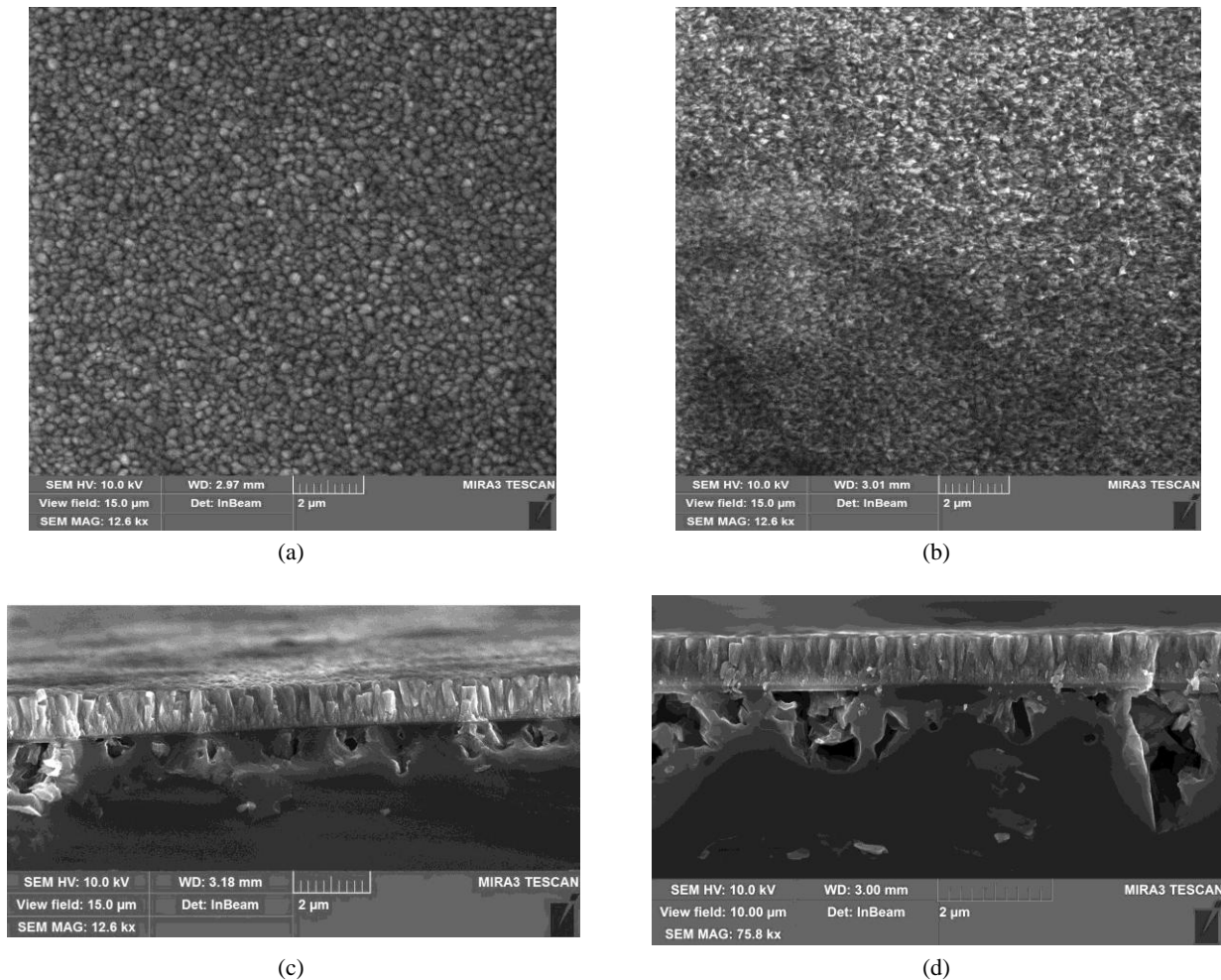
After formation of the SiC/porous Si/Si substrate, ZnO films were deposited onto the SiC surface. Deposition of ZnO films was performed by reactive high frequency magnetron sputtering of a zinc target. For deposition of ZnO films, the substrates were fixed using special clamps that carried out their movement inside the vacuum chamber. The substrates and target were placed in parallel to each other. The target for zinc oxide sputtering was a zinc disc of 80 mm in diameter and 6-mm thick, with 5N purity. During the growth of zinc oxide, argon (99.9995%) and oxygen (99.9995%) entered the operating chamber.

**Table 1.** Conditions for the synthesis of SiC layers on samples of porous Si/Si (100).

Parameter	Value
Temperature of growth, °C	1290
Time of growth, min	30
Total pressure of the CO + SiH <sub>4</sub> gas mixture, Pa	133
CO gas flow, cm <sup>3</sup> /min	14
SiH <sub>4</sub> gas flow, cm <sup>3</sup> /min	3.5

**Table 2.** Parameters of the process of ZnO films deposition.

Parameter	Value
Residual pressure in the vacuum chamber, Pa	10 <sup>-3</sup>
Pressure of oxygen P <sub>O<sub>2</sub></sub> , Pa	0.1
Pressure of argon P <sub>Ar</sub> , Pa	1
Temperature of substrate, °C	300
Power on the discharge of the RF magnetron, W	200
Distance target-substrate, m	0.07
Time of deposition, s	1200



**Fig. 1.** SEM images of the surface (a, c) and cleavage (b, d) of ZnO films grown on SiC/porous Si/Si substrates prepared at the oxygen pressure 0.1 and 0.6 Pa, respectively.

The parameters of the process of deposition of ZnO films by the method of magnetron sputtering of a zinc target are shown in Table 2.

### 3. Results and discussion

#### 3.1. Morphological studies

Fig. 1 shows the SEM images of the surface and cleavage of ZnO films grown on SiC/porous Si/Si substrates prepared at the oxygen pressure values 0.1 and 0.6 Pa.

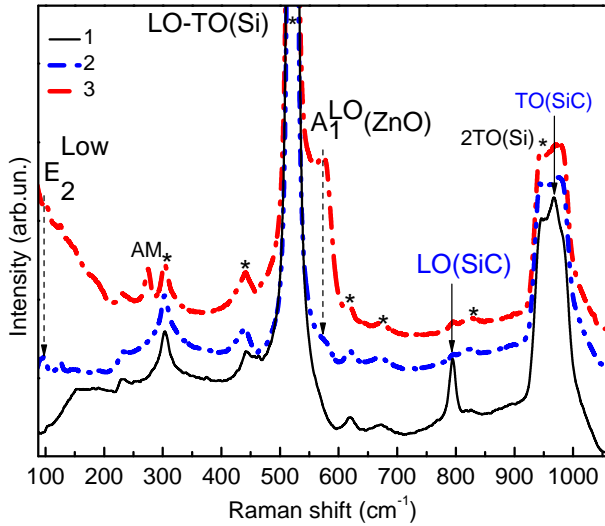
As can be seen in Fig. 1, the ZnO film prepared at the oxygen pressure 0.6 Pa has a smoother surface than those at the pressure 0.1 Pa. In both cases, a columnar structure typical for ZnO films is observed on the surface of the samples. In this case, the high packing density leads to coalescence of individual crystallites. With increasing the pressure, the thickness of the ZnO film somewhat decreases from 1108 nm at 0.1 Pa to 864 nm at 0.6 Pa. The decrease of the thickness of ZnO film is caused by a decrease in the film deposition rate with an increase in the partial pressure of oxygen, which is typical for the method of magnetron sputtering of films [14].

#### 3.2. Raman spectroscopy

Raman scattering (RS) was used to study the structural perfection and phonon properties of ZnO films deposited by magnetron sputtering on SiC/porous Si/Si substrates.

According to the group-theoretical analysis, optical phonons of ZnO are described by irreducible representations at the center of the Brillouin zone:  $\Gamma_{opt} = A_1 + 2B_1 + E_1 + 2E_2$  [15]. Polar  $A_1$  and  $E_1$  phonon vibrations are split into longitudinal optical (LO) and transverse optical (TO) phonon modes that manifest themselves in Raman and infrared (IR) spectra. Inelastic scattering by  $E_2$  (high) and  $E_2$  (low) phonon vibrations is allowed in Raman spectra, while the scattering by vibrations of  $B_1$  symmetry (silent modes) is forbidden and does not manifest in the optical spectra of structurally perfect crystals.

Fig. 2 shows the Raman spectra of ZnO films grown on SiC/porous Si/Si substrates under different oxygen partial pressures. The Raman spectra of all the samples contain Raman bands marked with the symbol \* at 303, 442, 521, 620, 671, 826, and 944 ... 980  $\text{cm}^{-1}$ , corresponding to the Raman bands of the first and second



**Fig. 2.** Raman spectra of the SiC/porous Si/Si substrate (1) and ZnO films grown under the pressure 0.1 Pa (2) and 0.6 Pa (3) on a SiC/porous Si/Si substrate. The symbol \* denotes the Raman bands of the Si substrate ( $\lambda_{exc} = 532.0$  nm,  $T = 300$  K).

orders in the Si substrate [16]. The Raman bands at 795 and 967  $\text{cm}^{-1}$  are caused by scattering related to LO and TO phonons of the layer of cubic 3C-SiC modification, respectively [17].

The Raman spectra of ZnO/SiC/porous Si/Si (Fig. 2, curves 2 and 3) measure a band at 574  $\text{cm}^{-1}$  corresponding to the scattering by  $A_1(\text{LO})$  phonons of ZnO. The relatively high intensity of the band in the non-resonant Raman spectra of ZnO films can be explained as a result of the defect-induced inelastic scattering. With an increase in the concentration of structural defects, an increase in the structural disordering of the ZnO crystal lattice takes place, which manifests itself in the Raman spectra in an increase in the intensity of polar  $A_1(\text{LO})$  mode and the practical absence of non-polar  $E_2$  (low) and  $E_2$  (high) phonon modes (Fig. 2). As it was shown by the authors of [18], the appearance of an ensemble of randomly oriented nanowires or columns that are characteristic of the ZnO structure (see Fig. 1) can lead to an increase in the intensity of polar  $A_1(\text{LO})$  mode. In this case, the increase in the intensity of the  $A_1(\text{LO})$  mode with respect to the non-polar  $E_2(\text{low})$  and  $E_2(\text{high})$  phonon modes is due to such effects as deformation and disordering in the ensemble of ZnO nanowires. With an increase in the partial pressure of oxygen, the intensity of the  $A_1(\text{LO})$  mode in the Raman spectra increases (Fig. 2). This may be due to the fact that with a decrease of the film thickness observed with increasing the partial oxygen pressure (Fig. 1), the contribution of disordering effects as well as the reflection of exciting radiation at the boundaries between the individual columns are increased.

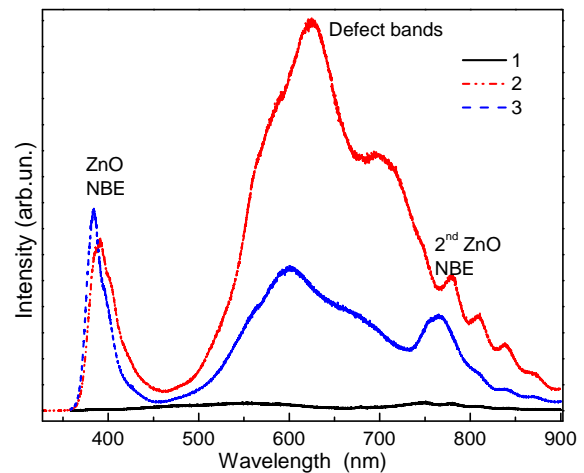
In the Raman spectrum of ZnO film grown under the oxygen pressure 0.6 Pa, an additional band is found at  $\sim 275.0$   $\text{cm}^{-1}$ , which cannot be associated with the process

of inelastic scattering of the first or second order by phonons of wurtzite ZnO. To date, the nature of this band remains unclear. In [19], a linear correlation was found between the nitrogen concentration and the intensity of vibrational bands at 275.0 and 582.0  $\text{cm}^{-1}$ , which allowed the authors to interpret these bands as local vibrations involving nitrogen. An increase in the intensity of the 275  $\text{cm}^{-1}$  band during thermal annealing [20] associated with vibrations of Zn atoms in the local environment, where some of the nearest neighboring oxygen atoms substitute nitrogen atoms in the ZnO lattice [21]. It is assumed in [22] that the additional modes can be caused by the effects of disordering of the ZnO crystal lattice and do not depend on the features of incorporation of nitrogen atoms into the ZnO matrix.

To confirm this assumption, they carried out polarization Raman studies of ZnO films doped with Fe, Sb, Al, Ga, and Li grown on a sapphire substrate by using the method of pulsed laser sputtering. Thus, additional modes at 277, 511, 583, and 644  $\text{cm}^{-1}$  were recorded in the Raman spectra of ZnO films doped with Fe, Sb, and Al. In [23], this additional band was assigned to a local vibrational mode in the Raman spectra of clusters formed by interstitial zinc  $\text{Zn}_i$ . In addition, the marked band may correspond to the  $B_1(\text{low})$  ZnO mode. A high level of impurities or structural defects leads to a weakening of the selection rules for the Raman scattering, and registration of this “forbidden” band becomes possible.

### 3.3. Photoluminescent research

Fig. 3 shows room-temperature PL spectra of ZnO films grown on a SiC/porous Si/Si substrate at various values of oxygen partial pressure, as well as the PL spectrum of the SiC/porous Si/Si substrate. The PL spectra of both structures contain intense emission bands, both in the ultraviolet and visible spectral ranges.



**Fig. 3.** PL spectra of the SiC/porous Si/Si substrate (1) and ZnO films grown under the pressure 0.1 Pa (2) and 0.6 Pa (3) on the SiC/porous Si/Si substrate.  $T = 300$  K,  $\lambda_{exc} = 325$  nm.

The PL spectrum of the ZnO film (Fig. 3, curves 2 and 3) with interband excitation  $\lambda_{\text{exc}} = 325 \text{ nm}$  ( $E_{\text{ex}} = 3.81 \text{ eV}$ ) contains an intense ultraviolet near-band-edge (NBE) emission band with a maximum at  $\sim 383 \text{ nm}$  for the case when the film was grown at the oxygen pressure 0.6 Pa and 388 nm for the case of the oxygen pressure 0.1 Pa, which is caused by the radiative recombination of free excitons [24]. The shape of this PL band has a low-energy asymmetry, which indicates the efficiency of radiative recombination processes involving longitudinal optical phonons [24]. The difference in the positions of the maxima inherent to the edge PL band in ZnO films at different pressures, as compared to bulk ZnO (368 nm) [25], is caused by the presence of elastic deformations through the mismatch between the lattice constants and thermal expansion coefficients of the film and substrate [26].

The broad emission band with the maxima at 560, 625 and 700 nm is associated with radiative processes that occur at deep energy levels in the band gap and are caused by various types of intrinsic defects, including ionized oxygen vacancies  $O_i$  [27, 28]. The decrease in the intensity of this emission band in the PL spectra of ZnO/SiC/porous Si/Si with an increase in the partial pressure of oxygen indicates a decrease in the concentration of intrinsic defects.

#### 4. Conclusions

Thus, as a result of the studies of ZnO films grown on the SiC/porous Si/Si substrate by applying the magnetron RF sputtering, it has been shown that ZnO films grown under the oxygen pressure 0.6 Pa are characterized by a smoother and more uniform surface than those obtained under the oxygen pressure 0.1 Pa. It has been found that the prepared ZnO films have the thickness close to 1108 nm (at 0.1 Pa) and 864 nm (at 0.6 Pa).

It has been shown that, with an increase in the oxygen partial pressure, the intensity of the  $A_1(\text{LO})$  mode ( $574 \text{ cm}^{-1}$ ) in the Raman spectrum of the ZnO/SiC/porous Si/Si structure increases, which indicates an increase in the structural disordering in the columnar structure of the ZnO film. At the same time, with an increase in the partial pressure of oxygen, the intensity of the broad emission band with the maxima at 560, 625, and 700 nm in the PL spectra of ZnO/SiC/porous Si/Si decreases, which indicates a decrease in the concentration of intrinsic defects, including ionized oxygen vacancies  $O_i$ .

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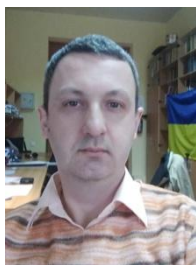
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**Karpenko O.Yu.:** sample preparation, manufacture of ZnO films.

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#### Формування плівок ZnO на підкладках SiC/пористий Si/Si

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**Анотація.** Методом реактивного магнетронного напилення одержано плівки ZnO на підкладках SiC/пористий Si/Si. При цьому плівка карбіду кремнію на поверхні пористого Si була отримана методом хімічного заміщення атомів. Показано, що плівки ZnO, вирощені при парціальному тиску кисню 0.6 Па, характеризуються більш гладкою та однорідною поверхнею, ніж покриття, отримані при тиску кисню 0.1 Па. На підставі аналізу спектрів раманівського розсіяння світла та фотолюмінесценції показано, що збільшення парціального тиску кисню приводить до збільшення структурного розупорядкування кристалічної ґратки ZnO з одного боку, і в той же час до зменшення концентрації власних дефектів, у тому числі іонізованих кисневих вакансій  $O_i$ , з іншого боку.

**Ключові слова:** реактивне магнетронне напилення, пористий Si, плівка карбіду кремнію, метод хімічного заміщення атомів, спектри раманівського розсіяння світла та фотолюмінесценції.